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## Key indicators

Single-crystal X-ray study
$T=294 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.006 \AA$
$R$ factor $=0.043$
$w R$ factor $=0.112$
Data-to-parameter ratio $=16.4$
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

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## 3,6-Dibromo-9-(4-tolylsulfonyl)-9H-carbazole

The title compound, $\mathrm{C}_{19} \mathrm{H}_{13} \mathrm{Br}_{2} \mathrm{NO}_{2} \mathrm{~S}$, was synthesized by N alkylation of 4-methylbenzenesulfonyl chloride with 3,6-dibromo- 9 H -carbazole. The carbazole ring system is essentially planar, with a mean deviation of $0.024 \AA$, and makes a dihedral angle of $75.46(9)^{\circ}$ with the plane of the benzene ring.

## Comment

Carbazole derivatives substituted by N -alkylation show useful pharmaceutical properties (Buu-Hoï \& Royer, 1950; Harfenist \& Joyner, 1983; Caulfield et al., 2002; Harper et al., 2002). This paper reports the structure of 3,6-dibromo-9-(4-tolylsulfonyl)9 H -carbazole, (I), which was synthesized by N -alkylation of 4methylbenzenesulfonyl chloride with 3,6-dibromo-9H-carbazole.


The carbazole ring system is essentially planar, with a mean deviation of $0.024 \AA$. The dihedral angle formed between the carbazole ring system and the plane of the benzene ring is 75.46 (1) ${ }^{\circ}$. The $\mathrm{C}-\mathrm{Br}$ distances are 1.915 (4) and 1.916 (4) A, consistent with literature values (Allen et al., 1987).

## Experimental

The title compound, (I), was prepared according to the procedure of Chakrabarti et al. (1989). 3,6-Dibromo-9H-carbazole (1.95 g) (Smith et al., 1992) in a solution of dimethylformamide ( 25 ml ) and benzene $(25 \mathrm{ml})$ was treated with sodium hydride $(0.168 \mathrm{~g})$ in an ice bath for 30 min . To the cold stirred solution, 4-methylbenzenesulfonyl chloride $(1.12 \mathrm{~g})$ was added and the mixture stirred at room temperature for a further 4 h . The resulting mixture was then poured into water ( 40 ml ) and extracted with benzene ( 100 ml ). After drying


Figure 1
A view of the molecular structure of (I). Displacement ellipsoids are drawn at the $30 \%$ probability level.
the benzene extracts over anhydrous sodium sulfate, filtration and concentration, the solid product was recrystallized from EtOH , giving crystals of (I) (yield: 1.36 g , $98 \%$; m.p. 494 K ). Compound (I) ( 40 mg ) was dissolved in a mixture of chloroform $(6 \mathrm{ml})$ and ethanol ( 2 ml ), and the solution was kept at room temperature for 10 d . Natural evaporation of the solution gave colourless crystals suitable for X-ray analysis.

## Crystal data

$\mathrm{C}_{19} \mathrm{H}_{13} \mathrm{Br}_{2} \mathrm{NO}_{2} \mathrm{~S}$
$M_{r}=479.18$
Monoclinic, $P 2_{1} / c$
$a=8.312$ (4) $\AA$
$b=20.573(11) \AA$
$c=11.374(6) \AA$
$\beta=106.867$ (9) ${ }^{\circ}$
$V=1861.3(16) \AA^{3}$

## Data collection

Bruker SMART CCD area-detector diffractometer
$\varphi$ and $\omega$ scans
Absorption correction: multi-scan (SADABS; Bruker, 1997)
$T_{\text {min }}=0.413, T_{\text {max }}=0.573$
(expected range $=0.385-0.534)$

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.043$
$w R\left(F^{2}\right)=0.113$
$S=1.04$
3733 reflections
227 parameters
H-atom parameters constrained

## $Z=4$

$D_{x}=1.710 \mathrm{Mg} \mathrm{m}^{-3}$
Mo $K \alpha$ radiation
$\mu=4.48 \mathrm{~mm}^{-1}$
$T=294$ (2) K
Block, colourless
$0.24 \times 0.20 \times 0.14 \mathrm{~mm}$

10105 measured reflections 3733 independent reflections 2508 reflections with $I>2 \sigma(I)$ $R_{\text {int }}=0.047$ $\theta_{\text {max }}=26.4^{\circ}$

$$
\begin{aligned}
& \begin{array}{c}
w=1 /\left[\sigma^{2}\left(F_{\mathrm{o}}^{2}\right)+(0.0552 P)^{2}\right. \\
\quad+0.1434 P] \\
\text { where } P=\left(F_{\mathrm{o}}^{2}+2 F_{\mathrm{c}}^{2}\right) / 3 \\
(\Delta / \sigma)_{\max }=0.003 \\
\Delta \rho_{\max }=0.67 \mathrm{e}^{-3} \AA^{-3} \\
\Delta \rho_{\min }=
\end{array}-0.65 \mathrm{e}^{-3}
\end{aligned}
$$



Figure 2
Part of the packing of the title compound, viewed down the $c$ axis. Dashed lines indicate hydrogen bonds.

Table 1
Hydrogen-bond geometry ( $\mathrm{A}^{\circ}{ }^{\circ}$ ).

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :--- | :--- | :--- | :--- | :--- |
| $\mathrm{C} 15-\mathrm{H} 15 \cdots \mathrm{O}^{\mathrm{i}}$ | 0.93 | 2.54 | $3.450(6)$ | 166 |

Symmetry code: (i) $x+1, y, z$.
All H atoms were included in the riding model approximation, with $\mathrm{C}-\mathrm{H}=0.93$ (aromatic) and $0.96 \AA$ (methyl), and with $U_{\text {iso }}(\mathrm{H})=$ $1.2 U_{\text {eq }}(\mathrm{C})$.

Data collection: SMART (Bruker, 1997); cell refinement: SAINT (Bruker, 1997); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1997); software used to prepare material for publication: SHELXTL.

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